organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Sandro Boland,^a Christine Cauvin,^a* Bernadette Norberg,^a Kiet Le Van,^b Benoît Georges,^b Laszlo Hevesi^b and François Durant^a

^aLab. Chimie Moléculaire Structurale, Facultés Universitaires N.-D. de la Paix, 61 Rue de Bruxelles, B-5000 Namur, Belgium, and ^bLab. Chimie Matériaux Organiques, Facultés Universitaires N.-D. de la Paix, 61 Rue de Bruxelles, B-5000 Namur, Belgium

Correspondence e-mail: christine.cauvin@fundp.ac.be

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.005 Å R factor = 0.059 wR factor = 0.185 Data-to-parameter ratio = 16.2

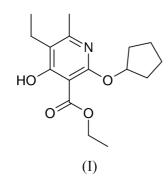
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Ethyl 2-(cyclopentyloxy)-5-ethyl-4-hydroxy-6-methylnicotinate

Structural analysis of the title compound, $C_{16}H_{23}NO_4$, reveals that the pyridine ring and the ester moiety are coplanar. This conformation is stabilized by an intramolecular hydrogen bond between the hydroxyl and ester groups.

Comment

In an attempt to design new HIV reverse transcriptase inhibitors (Parniak & Sluis-Cremer, 2000), the title compound (I), namely ethyl 2-(cyclopentyloxy)-5-ethyl-4-hydroxy-6methylnicotinate, was synthesized and studied by X-ray diffraction. The atom labeling and molecular conformation adopted for this compound are depicted in Fig. 1. The pyridine heterocycle is planar; the displacement of the atoms from their mean plane does not exceed 0.014 Å. The cyclopentyl ring is orthogonal to the pyridine heterocycle [dihedral angle of $89.59 (2)^{\circ}$ between the mean planes], with N1-C2-O10-C11 and C2-O10-C11-C12 torsion angles of -4.9 (4) and $-86.2 (3)^{\circ}$, respectively. Furthermore, the ester function is nearly coplanar with the pyridine ring [C4-C3-C16-C17 1.4 $(4)^{\circ}$]. This conformation is induced by an intramolecular hydrogen bond between the hydroxyl and ester moieties (O21-H21 \cdots O17, see Table 1). In addition to this hydrogen bond, the molecular structure of (I) is also stabilized by intermolecular $C-H \cdots O$ contacts (see Table 1).



Experimental

The title compound, (I), was synthesized from ethyl 5-ethyl-4hydroxy-6-methyl-pyridine-2(1H)-one-3-carboxylate (Dollé *et al.*, 1995). In a 25 ml two-necked flask under argon, the pyridinone (1.12 g, 5 mmol) and Ag₂CO₃ (0.717 g, 2.6 mmol) were dissolved in 10 ml benzene. The reaction mixture was warmed to 303 K and a slight excess of iodocyclopentane (640 µl, 5.5 mmol) was then added. After 18 h, the whole mixture was cooled to 273 K, diluted with 10 ml pentane, and filtered. This organic layer was washed successively with diluted NaHCO₃ and saturated NaCl, then dried with anhydrous Na₂SO₄, filtered off and evaporated to dryness. This sample was

 \odot 2002 International Union of Crystallography Printed in Great Britain – all rights reserved

Received 25 July 2002 Accepted 29 July 2002

Online 31 July 2002

purified on a SiO₂ column (eluant: dichloromethane/pentane: 3:1). 1.375 g (4.65 mmol, 93% yield) of a white solid were obtained. Spectroscopic analysis, ¹H NMR (90 MHz, CDCl₃, δ , p.p.m.): 1.1 (t, 3H, CH₃), 1.4 (t, 3H, CH₃), 1.5–2.2 (m, 8H), 2.4 (s, 3H, CH₃), 2.6 (q, 2H, CH₂), 4.3 (q, 2H, OCH₂), 5.5 (m, 1H, OCH), 12.5 (s, 1H, NH). Slow evaporation of a solution of (I) in ethanol gave colourless crystals suitable for X-ray analysis.

Z = 2

 $D_x = 1.215 \text{ Mg m}^{-3}$

Cell parameters from 25

Cu $K\alpha$ radiation

reflections

 $\theta = 30.0-40.0^{\circ}$ $\mu = 0.71 \text{ mm}^{-1}$

T = 293 (2) K

 $\begin{array}{l} R_{\rm int} = 0.022 \\ \theta_{\rm max} = 71.9^{\circ} \\ h = -5 \rightarrow 7 \end{array}$

 $k = -13 \rightarrow 13$

 $l = -14 \rightarrow 15$

3 standard reflections

frequency: 60 min

every 200 reflections

intensity decay: 3.5%

Platelet, colourless

 $0.21 \times 0.10 \times 0.09 \text{ mm}$

Crystal data

 $\begin{array}{l} {\rm C_{16}H_{23}NO_4}\\ {M_r} = 293.35\\ {\rm Triclinic,} \ P\overline{1}\\ a = 6.312 \ (1) \ {\rm \mathring{A}}\\ b = 10.571 \ (1) \ {\rm \mathring{A}}\\ c = 12.396 \ (2) \ {\rm \mathring{A}}\\ \alpha = 84.205 \ (7)^\circ\\ \beta = 77.098 \ (8)^\circ\\ \gamma = 88.827 \ (5)^\circ\\ V = 802.1 \ (2) \ {\rm \mathring{A}}^3 \end{array}$

Data collection

Enraf–Nonius CAD-4 diffractometer $\theta/2\theta$ scans Absorption correction: analytical (Alcock, 1970) $T_{min} = 0.865, T_{max} = 0.939$ 3449 measured reflections 3147 independent reflections 1678 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.076P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.059$	+ 0.272P
$wR(F^2) = 0.185$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.01	$(\Delta/\sigma)_{\rm max} = 0.023$
3147 reflections	$\Delta \rho_{\rm max} = 0.26 \text{ e } \text{\AA}^{-3}$
194 parameters	$\Delta \rho_{\rm min} = -0.19 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} O21 - H21 \cdots O17 \\ C19 - H19A \cdots O17^{i} \end{array}$	1.01	1.58	2.513 (3)	152
	0.97	2.83	3.211 (4)	105

Symmetry code: (i) 1 - x, -y, 2 - z.

The displacement parameter of the hydroxyl H atom was refined and the methyl groups were allowed to rotate about their local threefold axes.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1992); cell refinement: *CAD-4 EXPRESS*; data reduction: *HELENA* (Spek, 2000); program(s) used to solve structure: *SIR*97 (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL*97 (Sheldrick,

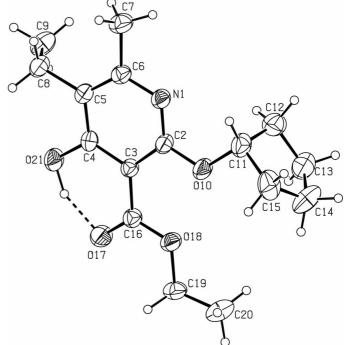


Figure 1

ORTEP view of compound (I), with displacement ellipsoids drawn at the 30% probability level.

1997); molecular graphics: *PLATON* (Spek, 2000); software used to prepare material for publication: *SHELXL*97.

SB and CC thank the Industry and Agriculture Research Foundation (FRIA) and the Région Wallonne (DGTRE) for financial support. The authors thank the Facultés Universitaires Notre-Dame de la Paix for the use of the Scientific Computing Facility.

References

- Alcock, N. W. (1970). Acta Cryst. A26, 437-439.
- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115–119.
- Dollé, V., Fan, E., Nguyen, C. H., Aubertin, A.-M., Kirn, A., Andreola, M. L., Jamieson, G., Tarrago-Litvak, L. & Bisagni, E. (1995). *J. Med. Chem.* 38, 4679–4686.
- Enraf-Nonius (1992). CAD-4 EXPRESS. Enraf-Nonius, Delft, The Netherlands.

Parniak, M. A. & Sluis-Cremer, N. (2000). Adv. Pharmacol. 49, 67-109.

- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2000). *HELENA* and *PLATON*. University of Utrecht, The Netherlands.